## **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**Cambridge International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2015 series

## 0654 CO-ORDINATED SCIENCES

**0654/61** Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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1 (a) (i) outline concave on one side and projections on the other; [2]

2 circles shaded and labelled;

(ii) xylem; [2]

transport of water;

(b)

test solution	observation	conclusion
Benedict's solution	orange	reducing sugar/glucose (present);
biuret solution	blue	protein absent ;
iodine solution	orange	starch absent ;

[3]

(c) Any 3 from 4 [3]

(celery in dyed water and) measure distance dye moves;

minimum 3 different temperatures;

time for coloured water to appear at top of (cut) stalk/set time and measure distance moved for each T;

all other conditions/named condition kept constant;

[Total: 10]

- **2** (a) 14 and 16; [1]
  - **(b) (i)** 0.7(0) 0.8(0); [3]

0.49 and 0.64;

 $T^2$  to 2 d.p.;

Allow ecf

- (ii) 4 plots correct  $\pm$  1/2 small square ; [2] best fit straight line through origin  $\pm$  1/2 small square ;
- (iii) gradient shown clearly on graph (triangle at least 1/2 of graph); [2] 1.6;
- (iv) 39.5/gradient from (b)(iii) = 25; quoted to 2 sig figs; [2]

[Total: 10]

Page 3		3		Syllabus	Paper					
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3	(a)	(i)	blue/pur		[1]					
		(ii)	calcium h		[2]					
			calcium o							
	(b)	(i)	(sodium		[3]					
			(ammoni	(ammonia) (light) blue ppt ;						
			(ammoni							
		(ii)	CuO (not	CuO (not name) ;						
							[3]			
	(c)	add	d sodium h	ct with (e.g.) sulphuric acid ; sodium hydroxide (soln)/ammonia (soln) ;						
		wni	ite ppt (als	solves in excess);			FT - 4 - 1 - 401			
							[Total: 10]			
4	(a)	(i)		e blood cell ;			[4]			
			<b>C</b> plate							
		<b>/::</b> \	<b>D</b> plasi		F41					
		(ii)					[1] [2]			
		(iii)		0.008;;						
			ecf							
	(b) (i)					_				
			activity	average pulse rate for 15 seconds	average heart rate (beats per minute)					
			resting	17	68					
			jogging	35	140					
					[1]					
		(ii)	heart rate		[max 1]					
			increased							
			need mo							
		(iii)	average	peats ;	[1]					
							[Total: 10]			

**Mark Scheme** 

Syllabus

**Paper** 

Page 3

Pa	ige 4	4			rk Scher		0045		Syllabus	Paper
			Camb	ridge IG	CSE – M	lay/June	2015		0654	61
5	5 (a)	use of cell/battery/power supply and connections;						[3]		
		connect in cir	connect in circuit ;							
		(first two mark	(first two marks can be from a diagram)							
		lamp works if	lamp works if lamp lights ;							
(b)	(b)	ammeter sym	ammeter symbol correct and in series with lamp;							[3]
		voltmeter sym	voltmeter symbol correct and in parallel with lamp ;							
		circuit;								
	(c)									[3]
		(lamp)	eg A	В	С	D	E			
		current/A								
		potential difference/V								
		table with hea	table with headings (allow p.d.) ;							
			correct units (allow name or symbol); room for 5 lamps may be labelled with letters, numbers or not at all;							
	(d)	resistance = potential difference (voltage)/current;						[1]		
										[Total: 10]
6	(a)	hydrogen; lighted splint pop (etc.);	;							[3]
	(b)	conical flask with delivery tube; (connected to) syringe or measuring cylinder over water;							[2]	
	(c)	(i) rate decreases; (then) stops;							[2]	
		(ii) Mg or acid or reactant(s) used up/all Mg or acid or reactant reacted;					d ;	[1]		
	(d)	line <b>T</b> to left o								[2]
										[Total: 10]